



Europäisches Patentamt  
European Patent Office  
Office européen des brevets



Publication number: **0 455 307 A1**

12

## EUROPEAN PATENT APPLICATION

21 Application number: **91201047.7**

51 Int. Cl.<sup>5</sup>: **B01J 23/00, B01J 37/00,  
C04B 35/48, C04B 35/44**

22 Date of filing: **02.05.91**

30 Priority: **04.05.90 GB 9010075**

43 Date of publication of application:  
**06.11.91 Bulletin 91/45**

84 Designated Contracting States:  
**AT BE CH DE DK ES FR GB GR IT LI LU NL SE**

71 Applicant: **SHELL INTERNATIONALE  
RESEARCH MAATSCHAPPIJ B.V.**  
Carel van Bylandtlaan 30  
NL-2596 HR Den Haag(NL)

72 Inventor: **Joustra, Annie Hendrika**  
Badhuisweg 3  
NL-1031 CM Amsterdam(NL)  
Inventor: **Scheffer, Bob**  
Badhuisweg 3  
NL-1031 CM Amsterdam(NL)

74 Representative: **Spierenburg, Jan et al**  
Shell Internationale Research Maatschappij  
B.V. Patents, Licensing & Trade Marks  
Division P.O. Box 302  
NL-2501 CH 's-Gravenhage(NL)

54 Process for the preparation of alumina-based extrudates.

57 A process for the preparation of an alumina-based extrudate suitable for use in the manufacture of a catalyst, comprises the steps of:

- i) preparing an extrudable mixture comprising an alumina precursor, a cobalt, iron or nickel source and a solvent;
- ii) extruding the mixture; and
- iii) drying the so-formed extrudate.

The extrudable mixture may optionally comprise a promoter source. Alternatively, a promoter source may be deposited on the resulting extrudate. A catalyst is prepared by calcination of the resulting extrudate, which catalyst is useful in the preparation of hydrocarbons from methanol or a mixture of carbon monoxide and hydrogen.

EP 0 455 307 A1

The present invention relates to a process for the preparation of an alumina-based extrudate suitable for use in the manufacture of a catalyst, to the extrudate so-obtained, to a process for the manufacture of a catalyst using this extrudate, and to a catalytic process in which the alumina-based extrudate catalyst is used.

5 The preparation of hydrocarbons from mixtures comprising hydrogen and carbon monoxide by contacting this mixture at elevated temperature and pressure with a catalyst is known in the literature as the Fischer-Tropsch synthesis. A catalyst often used for this hydrocarbon synthesis comprises cobalt, iron and/or nickel, a promoter, and a carrier material. These catalysts can suitably be prepared by deposition of the active components on the carrier material techniques known in the art, such as precipitation, impregna-  
10 tion, kneading and melting.

Recently a class of Fischer-Tropsch catalysts has been found, which catalysts have the property of yielding a hydrocarbon mixture comprising only very few olefins and oxygen-containing compounds and consisting substantially of unbranched paraffins. A considerable portion of the hydrocarbons in this mixture boil above the boiling temperature of kerosine and gas oil fractions obtained in the conventional at-  
15 mospheric distillation of crude mineral oil (the so-called middle distillates; boiling range 150-360 °C). It has been found that the high boiling portion of this product may be converted at high yields into middle distillates by means of hydrocracking. As a feedstock for hydrocracking at least a portion of the product is chosen such that the initial boiling point lies above the final boiling point of the heaviest middle distillates desired as end product. Hydrocracking of the product, typically proceeding at very low hydrogen consump-  
20 tion, yields middle distillates having a considerably better pour point than those obtained by the direct synthesis of middle distillates using the Fischer-Tropsch hydrocarbon synthesis process.

Fischer-Tropsch catalysts in the aforementioned class typically comprise silica, alumina or silica/alumina as carrier material, and cobalt, iron or nickel, together with a promoter selected from the groups IIa, IIb, IVb and Vb of the Periodic Table, for instance zirconium, titanium, chromium or magnesium.  
25 During the preparation of these catalysts, the catalytically active metal, that is cobalt, iron or nickel, and the promoter are deposited on the carrier material, after which the resulting catalyst is calcined. The catalyst is activated before final use in the Fischer-Tropsch synthesis by means of reduction, the purpose of which is to reduce oxides of the catalytic species to their metallic states.

When alumina is used as the carrier material, a number of drawbacks have been encountered. Firstly, it  
30 has been observed that during the preparation of alumina-based catalysts the addition of ruthenium and/or platinum/palladium is required in order to increase the reducibility of the catalytic metal to an acceptable level. Even so, relatively high reduction temperatures are still required for such catalysts. The cost of these metals is very high and is prohibitive of the industrial production of these catalysts.

Secondly, it is known in the art that catalysts of the aforementioned class, particularly those comprising  
35 cobalt supported on alumina, show a lower carbon monoxide hydrogenation activity in comparison to similar catalysts comprising cobalt supported on silica (Journal of Catalysis Volume 85, page 78-88, 1984).

Most surprisingly, research has now revealed that the aforementioned drawbacks may be overcome if, in the manufacture of this type of catalyst, use is made of an alumina-based extrudate comprising a co-  
40 extruded cobalt, iron or nickel source. It is noted that the manner by which a promoter may be incorporated in the extrudate is not critical, and it may be co-extruded with the alumina and the cobalt, iron or nickel source or it may be deposited on the alumina-based extrudates by conventional means. The extrudates obtained show a good crush strength and are very suitable for the preparation of Fischer-Tropsch catalysts showing high activity, selectivity and stability. Furthermore, these catalysts can be easily reduced at reduction temperatures below 300 °C, preferably at 260 to 270 °C so that in situ reduction is industrially  
45 feasible.

Accordingly, the present invention provides a process for the preparation of an alumina-based extrudate suitable for use in the manufacture of a catalyst, comprising the steps of:

- i) preparing an extrudable mixture comprising an alumina precursor, a cobalt, iron or nickel source and a solvent;
- 50 ii) extruding the mixture; and
- iii) drying the so-formed extrudate.

In a preferred embodiment of the invention a cobalt source is used in the preparation of the extrudable mixture.

The matrix of the extrudate comprises an extruded alumina precursor. The alumina precursor may  
55 comprise aluminium oxides, aluminium hydroxides, aluminium oxide hydroxides, or mixtures thereof. Preferred alumina precursors are aluminium hydroxides having free hydroxyl groups and which, upon calcination, are converted into porous alumina of good crush strength. Examples of aluminium hydroxides are boehmite, pseudoboehmite, diaspor, gibbsite, bayerite and nordstrandite. Most preferred alumina

precursors are diaspor, boehmite and, in particular, pseudoboehmite. Upon calcination, diaspor is converted into alpha-alumina, whereas boehmite is converted into gamma-alumina, delta-alumina, or alpha-alumina, at calcination temperatures of 450 to 560 °C, 750 to 1000 °C and greater than 1000 °C, respectively. Pseudoboehmite tends to the formation of fibers on ageing and drying. The surface area of the alumina  
 5 may be as high as 500 m<sup>2</sup>/g.

The alumina precursor is mixed with the cobalt, iron or nickel source and the solvent to form an extrudable mixture. The solvent is used to provide an intimate mixture of the alumina precursor and the cobalt, iron or nickel source and may be any of the suitable solvents known in the art.

Solvents suitable for providing an extrudable mixture or dough include polar solvents, such as water and  
 10 lower alkanols, such as ethanol and methanol, and mixtures thereof. The solvent is most conveniently water. The amount of solvent used to form the extrudable mixture is such that the resulting mixture has the appropriate extrusion properties, that is the mixture may be easily extruded through the appropriate die plate with the resulting extrudates retaining their form during the drying procedure. Typically, the extrudable mixture comprises the solvent in an amount of from 20% by weight to 120% by weight on the basis of the  
 15 dry weight of the alumina precursor.

The extrudable mixture is preferably in the form of a dough, obtainable by the peptisation of the alumina precursor. Suitable peptising agents for inclusion in the extrudable mixture are well known in the art and include weak acids, for example formic acid, acetic acid and propionic acid. Acetic acid is a particularly preferred peptising agent. The amount of peptising agent is sufficient to form a dough and is typically from  
 20 1 to 25% by weight on the basis of the dry weight of the alumina precursor. Preferably, the extrudable mixture comprises from 3 to 20% by weight peptising agent. It has been found that extrudable mixtures comprising a peptising agent in an amount in the upper region of the preferred range, that is from 10 to 20% by weight, particularly about 15% by weight, result in particularly strong alumina-based extrudates.

Although alumina is preferably used as the extrudate matrix, mixtures of alumina with titania and/or  
 25 silica and/or zirconia may be used, provided that the extrusion properties of the mixture are not adversely affected.

An essential feature of the process according to the invention is the co-extrusion of the cobalt, iron or nickel source with the alumina precursor. The cobalt, iron or nickel source comprises any compound suitable for forming an extrudable mixture with the alumina precursor and the solvent and allowing the  
 30 cobalt, iron or nickel compound to be reduced to the metallic element. Examples of suitable cobalt, iron or nickel-containing compounds are inorganic or organic salts, for example nitrates, hydroxides and oxides. Specific examples for the sources for cobalt are cobalt nitrate and cobalt hydroxide. In general, hydroxides are preferred as sources for the cobalt, iron or nickel because during reduction the volatile decomposition product (water) does not result in environmental problems.

Generally, the amount of the cobalt, iron or nickel source is selected such that the extrudates prepared  
 35 comprise 10 to 300 parts by weight cobalt, iron or nickel per 100 parts by weight alumina. Preferably, the amount of cobalt, iron or nickel in the final extrudate lies in the range of about 25 to 200 parts by weight, preferably 40 to 100 parts by weight, most preferably 50 to 100 parts by weight per 100 parts by weight alumina.

The rheology of the extrudable mixture may be improved by adding a structuring agent, such as polyacrylamides, for example those sold under the trademark NALCO, and carboxymethyl cellulose, for  
 40 example sold under the trademark AVICEL.

To improve the flux properties of the extrudable mixture in the extruder a surface active agent or a polyelectrolyte may be added. The surface active agent or the polyelectrolyte lowers the viscosity of the  
 45 mixture, gives a smoother extrudate texture and facilitates cutting of the extruded product. Further, formation of macropores in the calcined catalytic material may be improved by the inclusion of such agents, which may in turn enhance the catalytic properties of the final products. Suitable surface active agents include cationic surface active agents, for example fatty amines, quaternary ammonium compounds, aliphatic monocarboxylic acids, ethoxylated alkyl amines, polyvinyl pyridine, sulfoxonium, sulfonium,  
 50 phosphonium and iodonium compounds, anionic surface active agents, for example alkylated aromatics, acyclic monocarboxylic acids, fatty acids, sulfonated aromatics, alcohol sulfates, ether alcohol sulfates, sulfated fats and oils and phosphonic acid salts and non-ionic surface active agents, for example polyoxyethylene alkylphenols, polyoxyethylene alkylamides, polyols and acetylenic glycols.

The extrudable alumina precursor, the cobalt, iron or nickel source and the solvent are co-mulled such  
 55 that by the intimate mixing of these ingredients the extrudable mixture is formed. The mixing time is typically from about 10 to 100 minutes, preferably about 35 to 80 minutes. Generally, the extrudable mixture has a solids content of from about 15 to 65% by weight, preferably about 25 to 55% by weight.

After co-mulling, the extrudable mixture may be extruded using a conventional extruder. An extruder of

the screw-type may be used to extrude the mixture through a die plate with orifices of the desired form and size. The extruded strands so-formed are cut to the desired length and then dried. The extrudates may be formed with any appropriate cross-sectional design, for example, circular, multi-lobed forms or the extrudate forms disclosed in US patent 4,028,227. Trilobe extrudates and hollow cylinders are very suitable products.

5 The extrudates may comprise one or more promoters selected from the groups IIa, IIIb, IVb and Vb of the Periodic Table, for example zirconium, titanium, chromium or magnesium, in particular zirconium. Rhenium may also be used as a promoter. The promoter may be added to the extrudate by conventional deposition techniques, or by incorporation in the mixture to be extruded. If the promoter is to be deposited on the extrudates, any conventional deposition technique may be used, such as impregnation. Conventional  
10 deposition methods are disclosed in US patent 4,522,939. Preferably, the promoter is added to the extrudable mixture, so that the use of a separate deposition step may be avoided and the required extrudates may be formed in one single operation.

Any promoter source may be used. When the promoter source is added to the extrudable mixture, it is desired that the extrusion properties of the mixture are not adversely affected. Both inorganic and organic  
15 promoter compounds are suitable for incorporation in the extrudable mixture. Examples of suitable compounds are oxides, hydroxides, and carbonates. Suitable zirconium compounds are zirconium hydroxide, zirconium carbonate and ammonium zirconium carbonate.

Although the amount of promoter used depends on the manner by which the promoter is added (deposition or co-extrusion), the amount of promoter generally lies in the range of about 0.1 to 100 parts by  
20 weight promoter per 100 parts by weight alumina. Preferably, the amount of promoter is 5 to 50 parts by weight alumina.

In a further aspect, the present invention provides alumina-based extrudates obtainable by the process as hereinbefore described.

The present invention further provides a process for the preparation of a catalyst comprising calcining  
25 the alumina-based extrudate prepared by the process hereinbefore described. The calcination temperature is typically in the range of from about 300 to 1000 °C, preferably about 400 to 900 °C. If the extrudable mixture used for forming the extrudate does not comprise a promoter, the process for the preparation of the catalyst optionally comprises an additional step in which the promoter compound is deposited on the extrudate prior to calcination.

30 After calcination the catalyst is activated by reduction, preferably at a reduction temperature of less than 300 °C. Most preferred is a reduction temperature in the range of about 250 to 280 °C. This reduction may be in situ, that is performed with the catalyst retained in the reactor. The reduction is very suitably carried out by contacting the catalyst with hydrogen or a hydrogen-containing gas.

In addition, the present invention provides a catalyst obtainable by the process hereinbefore described.

35 In yet a further aspect, the present invention provides a process for the synthesis of hydrocarbons using the aforementioned catalyst.

The preparation of hydrocarbons from a feed comprising synthesis gas, optionally in combination with methanol, comprises contacting the feed with the above described catalyst and is generally performed at a temperature of from 100 to 600 °C, more typically 150 to 350 °C, preferably 180 to 270 °C, most  
40 preferably 200 to 250 °C, and at a total pressure of generally about 1 to 200 bar absolute, preferably 10 to 70 bar absolute. The space velocity is typically from about 200 to 20,000 m<sup>3</sup> (STP) gaseous feed/m<sup>3</sup> reaction zone/hour. The term "STP" as used herein refers to a Standard Temperature (of 0 °C) and Pressure (of 1 bar absolute). Preferably, the feed comprises hydrogen and carbon monoxide in a molar ratio of 0.9:1 to 3:1, preferably 1:1 to 2.3:1.

45 In a preferred embodiment of the process according to the present invention, the preparation of hydrocarbons from carbon monoxide and hydrogen is used as the first step in a two-step process for the preparation of middle distillates.

To this end the hydrocarbon product, or at least that part of the product which has an initial boiling point above the final boiling point of the desired middle distillate fraction, is subjected to a catalytic hydrotreat-  
50 ment as the second step in the process. The catalytic hydrotreatment is suitably carried out by contacting the hydrocarbon material from the first step at elevated temperatures and pressures and in the presence of hydrogen with a catalyst comprising one or more metals having hydrogenation activity, supported on a carrier.

In the hydrotreatment preference is given to the use of a catalyst comprising one or more metals from  
55 Group VIII, supported on a carrier. In particular, a catalyst is preferred comprising platinum on a carrier, 13 to 15% by weight of which is alumina and the remainder of silica. Preferred reaction conditions in the hydrotreatment are temperatures in the range of from about 175 to 400 °C, in particular in the range of from 250 to 350 °C, a hydrogen partial pressure of 1 to 25 MPa, in particular 2.5 to 15 MPa, a space

velocity of 0.1 to 5 kg./h, in particular 0.25 to 2 kg./h and a hydrogen/oil ratio of 100 to 5000 NI/kg, in particular 250 to 2500 NI/kg.

The process for the preparation of the alumina-based extrudates and for the preparation of catalysts using the extrudates so-obtained is further described in the following illustrative examples.

5

#### EXAMPLE 1

##### Preparation of the extrudate

10 Various extrudates according to the invention, the composition of which are given in Table I, were prepared as follows:

Pseudoboehmite was peptised using diluted glacial acetic acid (3% by weight on dry basis). Thereafter the cobalt source and zirconium source were added in any desired order. Finally, water was added in such an amount that for the extrusion mixture the required loss on ignition (LOI) was obtained. This mixture was  
15 co-mulled by intimate mixing for the given mixing time. Finally, the structuring agent, AVICEL, was added and optionally a flux improving agent.

The extrudable mixture obtained was extruded using a 1" Bonnet extruder with a 1.7 mm delrin trilobe matrix. The extrudates, having a trilobe form, were dried and calcined at about 500 °C.

All extrusion mixtures showed a good to excellent extrudability.

20

##### Catalytic performance

The calcined extrudates, now called catalysts and indicated with the same numeral, were crushed, loaded into a reactor and dried at 260 °C in a nitrogen stream. The dried catalyst was reduced by  
25 increasing the hydrogen content of the nitrogen stream. The reduction temperature was 260 °C.

After reduction the temperature was lowered to about 195 °C and the pressure was gradually increased to 26 bar.

Synthesis gas was introduced into the reactor at a gas hourly space velocity of 800 NI/hr. The temperature of the catalyst was adjusted such that a H<sub>2</sub> + CO conversion of about 60 to 70% was obtained, corresponding to a space time yield (STY) of about 100 g/l/h. The performance of these catalysts,  
30 expressed as C<sub>5</sub> + selectivity is given in Table II.

For comparison purposes the catalytic performance of two prior art catalysts was tested, namely an impregnated 25Co/1Zr/100SiO<sub>2</sub> catalyst and an impregnated 25Co/1La/0.5Ru/100Al<sub>2</sub>O<sub>3</sub> catalyst. The former catalyst was reduced at 260 °C, whereas the latter catalyst was reduced at 350 °C. Under similar synthesis  
35 conditions the C<sub>5</sub> + selectivity was 90% and 88% by weight, respectively. Thus, it can be seen that the catalysts prepared according to the present invention suffered neither of the disadvantages associated with the prior art alumina-based catalysts discussed above and, indeed, have shown improved activity.

40

45

50

55

TABLE I

Material	Extrudate			
	1	2	3	4
Alumina precursor <sup>1)</sup> :				
Pseudoboehmite	1337	84	84	76
Co-source <sup>1)</sup> :				
Co(OH) <sub>2</sub>	1264	28	28	72
Zr-source <sup>1)</sup> :				
Zr(OH) <sub>4</sub>	580	13	--	--
(NH <sub>4</sub> ) <sub>2</sub> Zr(CO <sub>3</sub> ) <sub>2</sub>	--	--	14	33
Solvent (g):				
water	1275	80	70	20
Acetic acid (g):	87	4	4	5
Structuring agent (g):	40	1.6	1.6	2.3
LOI (% by wt):	47	50	53	51
Mixing time (min.):	40	57	70	57
Extrudability:	excellent	good	smooth extrudate	smooth extrudate
Weight ratio				
Al <sub>2</sub> O <sub>3</sub> /Co/Zr:	100/70/32	100/25/12	100/25/12	100/70/32

1) dry weight (g) on the basis of oxides

TABLE II

		Catalyst			
	Conditions	1	2	3	4
	Reduction				
5	temperature (°C)	260	260	260	260
10	H <sub>2</sub> /CO feed ratio	1.1	1.1	1.1	2.0
15	Temperature (°C)	202	220	222	193
	Run (hr)	135	100	110	70
20	H <sub>2</sub> +CO conversion	67	61	65	66
	STY (g/l/h)	108	100	105	106
25	C <sub>5</sub> + selectivity				
	(% wt)	91	87	89	89

## EXAMPLE 2

Six further extrudates were prepared using the basic procedure outlined in Example 1 above. The compositions of the extrudates are set out in Table III.

The crush strength (N/m) of the extrudates obtained was tested, the values of which are given in Table III. It can be seen that the process of the present invention can be used to prepare alumina-based extrudates having a very high crush strength. In particular, it can be seen that the use of amounts of peptising agent in the upper region of the preferred 3 to 20% range gives rise to extrudates having a particularly high crush strength.

TABLE III

Material	Extrudate					
	5	6	7	8	9	10
Alumina precursor <sup>1)</sup> : Pseudoboehmite	75	75	75	75	75	75
Co-source <sup>1)</sup> : Co(OH) <sub>2</sub>	71.5	71.5	71.5	71.5	71.5	71.5
Zr-source <sup>1)</sup> : Zr(OH) <sub>4</sub>	32.4	32.4	32.4	32.4	32.4	32.4
Solvent (g): water	100	100	100	80	80	80
Acetic acid (% wt) <sup>2)</sup> :	6.5	6.5	6.5	15	15	15

1) dry weight (g) on the basis of oxides

2) on the basis of dry weight of alumina



TABLE III (Cont'd)

	Extrudate					
	5	6	7	8	9	10
Material						
Structuring agent (% wt):	3.6	3.6	3.6	3.6	3.6	3.6
LOI (% by wt):	47.5	47.5	47.5	45.8	45.8	45.8
Mixing time (min.):	74	74	74	80	80	80
Extrudability:	Excellent	Excellent	Excellent	Excellent	Excellent	Excellent
Calcining temperature (°C):	560	630	700	560	630	700
Crush Strength (N/m):	51	53	45	74	78	89

## Claims

1. Process for the preparation of an alumina-based extrudate suitable for use in the manufacture of a catalyst comprising the steps of:
  - i) preparing an extrudable mixture of an alumina precursor, a cobalt, iron or nickel source and a

solvent;

ii) extruding the mixture; and

iii) drying the so-formed extrudate.

- 5    2. Process as claimed in claim 1, characterized in that the alumina precursor is selected from the group comprising boehmite, pseudoboehmite, gibbsite and mixtures thereof.
3. Process as claimed in either of claims 1 or 2, characterized in that the cobalt, iron or nickel source is a compound selected from the group comprising cobalt, iron or nickel carbonate, hydroxide, nitrate and mixtures thereof.
- 10    4. Process as claimed in any preceding claim, characterized in that the extrudable mixture comprises a cobalt source.
- 15    5. Process as claimed in any preceding claim, characterized in that the extrudable mixture further comprises a peptising agent.
6. Process as claimed in claim 5, characterized in that as peptising agent the mixture comprises from 10 to 20% by weight acetic acid on the basis of dry weight alumina.
- 20    7. Process as claimed in any preceding claim, characterized in that the extrudable mixture further comprises a promoter source wherein the promoter is selected from the groups IIa, IIIb, IVb and Vb of the Periodic Table.
- 25    8. Process as claimed in claim 7, characterized in that the promoter source is a compound selected from hydroxides, carbonate and carbonates of the promoter.
9. Process for the preparation of a catalyst comprising calcining an alumina-based extrudate prepared by a process as claimed in any one of claims 1 to 8.
- 30    10. Process for the preparation of a catalyst comprising the steps of:
  - i) depositing on an alumina-based extrudate prepared by a process as claimed in any one of claims 1 to 6 a promoter source, wherein the promoter is selected from the groups IIa, IIIb, IVb and Vb of the Periodic Table; and
  - 35    ii) calcining the extrudate.
11. An extrudate obtainable by a process as claimed in any one of claims 1 to 8.
12. A catalyst obtainable by a process as claimed in either of claims 9 or 10.
- 40    13. Process for the preparation of hydrocarbons comprising contacting methanol and/or a mixture of carbon monoxide and hydrogen with a catalyst as claimed in claim 12.

45

50

55



European  
Patent Office

## EUROPEAN SEARCH REPORT

Application Number

EP 91 20 1047

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int. Cl.5)
X,Y	CHEMICAL ABSTRACTS, vol. 114, no. 22, June 3, 1991, Columbus, Ohio, US; abstract no. 210402F, 'Preparation of hydrotreating extruded catalyst for petroleum fractions ' * abstract & CS-B-264 792 (J.VYBIHAL ET AL.) December 15, 1989 * - - -	1-6,9,7-8, 10-12	B 01 J 23/00 B 01 J 37/00 C 04 B 35/48 C 04 B 35/44
Y	NN-A- (Latest)ACCESSION NUMBER 89-197055 (DW 8927) DERWENT PUBLICATIONS LTD.,LONDON,GB & JP-A-1135530 (MITSUBISHI ET AL.)(29-5-1989) - - -	7-8,10-12	
A	US-A-4 388 222 (G.WILSON ET AL.) * the whole document * - - -	1-13	
A	EP-A-0 199 399 (SHELL) * the whole document * - - -	1-13	
A	NL-A-7 614 173 (UOP INC.) - - - - -		
			TECHNICAL FIELDS SEARCHED (Int. Cl.5)
			B 01 J C 04 B
The present search report has been drawn up for all claims			
Place of search		Date of completion of search	Examiner
The Hague		09 August 91	HARBON J.L.
<div>CATEGORY OF CITED DOCUMENTS</div> <div>X: particularly relevant if taken alone Y: particularly relevant if combined with another document of the same category A: technological background O: non-written disclosure P: intermediate document T: theory or principle underlying the invention</div> <div>E: earlier patent document, but published on, or after the filing date D: document cited in the application L: document cited for other reasons ----- &amp;: member of the same patent family, corresponding document</div>			

**THIS PAGE BLANK (USPTO)**